Subject: Biochemistry Year and Sem: F. Y. B. Pharm Sem II

Duration: 3 Hrs Total marks: 80 M

N.B.: 1. All questions are compulsory

2. Figures to right indicate full marks

Q. 1	Choose appropriate option for following multiple choice-based questions.	20
1	Which of the following is an aliphatic amino acid with R group containing	2,42
•	sulfhydryl group?	3,60
a	Phenyl alanine	55
<u>u</u>	lysine	875
c	Threonine	0 0 0 1 0 0 1 0 0
d	cysteine	35
		×,
2	The process of change in the specific optical rotation representing the interconversion of α and β forms of D-glucose to an equilibrium mixture is called as	
a	Mutarotation	
b	Epimerization	
С	Racemization	
d	Inversion	
3	Glucose and mannose are examples of	
a	C4 epimers	
b	C2 epimers	
c	C6 epimers	
d	C1 epimers	
4	The conversion of phosphoenol pyruvate to pyruvate catalysed by enzyme	
•	pyruvate kinase resulting in the synthesis of ATP is an example of	
a	Oxidative Phosphorylation	
b	Oxidative dephosphorylation	
C C	Substrate level Phosphorylation	
o d	Photophosphorylation	
25 S	\$\frac{1}{2}\langle \frac{1}{2}\langle \frac{1}{2}\	
5	is useful in generating pentoses and NADPH, required for the	
	biosynthetic reactions.	
a	HMP shunt	
b	Glycolysis	+
) C	TCA cycle	
d	Gluconeogenesis	
	Oluconcogenesis (
6	The total ATP yield from oxidation of one mole of acetyl CoA by TCA cycle is	
a	80000000000000000000000000000000000000	
b		1
S C	24 6 6 6 6 6 6 6 6 6 6 6 6 6 6 6 6 6 6 6	
d	16	

7	Which of the following enzyme is common for gluconeogenesis and glycolysis?
a	Glyceraldehyde-3-phosphate dehydrogenase
<u>u</u> b	Hexokinase
c	Pyruvate kinase
d	Pyruvate carboxylase
<u>u</u>	1 yluvute eurooxyluse
0	
8	The reducing equivalents are supplied byin fatty acid biosynthesis.
a	NADPH
b	NADH SEES SAN SEES SA
С	NAD STATE OF
d	FADH ₂
9	The regulatory step in the synthesis of AMP from IMP is catalyzed by
a	Glutamine-phosphoribosyl pyrophosphate amido transferase
<u>a</u> b	Adenylosuccinate synthetase
c	Adenylosuccinate lyase
d	IMP dehydrogenase
u	IIVII denydrogenase
40	
10	separates the two strands of DNA during replication.
<u>a</u>	Gyrase
b	Topoisomerase
C	Helicase DNA
d	DNA polymerase
11	Which of the following work is done by the sigma factor in transcription?
<u>a</u>	Helicase action
<u>u</u>	Transcription initiation
c	Transcription elongation
d o	Transcription termination
2016	
12	Fluorouracil inhibits the activity of
a	Dihydrofolate reductase
b	Thymidylate synthase
5 5 C	CTP synthase
d	Ribonucleotide reductase
ST.	Ationacional feduciase
796	
13	is involved in salvage pathway of purines.
o a	Adenine phosphoribosyl transferase
b	Glutamine- PRPP amidotransferase
C S	IMP dehydrogenase
od)	Uridine-cytidine kinase

14	The number of ATP molecules formed by β -oxidation of one mole of palmitic	
	acid are	
a	126	2 0 6 6 6 6 6 6 6 6 6 6 6 6 6 6 6 6 6 6
b	106	
c	135	
d	108	2 2 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0
u		2222
15	of the following is the regulatory step of cholesterol biosynthesis.	7 12 12 12
a	Formation 3-hydroxy-3-methylglutaryl CoA	3000000
b	Formation of Mevalonate	5555
C	Formation of Isoprenoid Unit	8.5.5.
d	Formation of acetoacetyl CoA	16,67
<u>u</u>		50,
16	The accumulation of acetyl CoA in the mitochondria of the liver results in	×.
10	generation of	
a	ATP	
b	Ketone bodies	
c	Free fatty acids	
d	Oxaloacetate Oxaloacetate	
17	is involved in biosynthesis of dopamine.	
17 a	Tyrosine hydroxylase	
b	Tyrosinase	
c	Phenylethanolamine N-methyltransferase	
d	Dopamine β-hydroxylase	
u	Doparinic p hydroxytase	
18	is the link between urea cycle and TCA cycle.	
a	Fumarate	
b o	Succinate	
C	α- ketoglutarate	
& d	Citrate	
19.39.5		
19	catalyses the rearrangement reactions involving atomic grouping	
	without altering molecular weight or number of atoms.	
8 8 a	Ligase	
b	Isomerase	
TO C	Oxidoreductase	
o √ do	Hydrolase	
6, 93 ° 74, 74, 74		
20	The functional unit of the enzyme is known as	
a	Chiroenzyme Company Co	
b	Holoenzyme	
	Prosthetic group	
To d	Monomeric enzyme	
20 27 29 06 20 27 29 06		

2. 2 A	Answer any one question.	12
a	 i) Write a note on Carnitine shuttle. ii) Explain the biosynthesis of noradrenaline with its significance. iii) Give the reactions catalysed by FAS complex in the biosynthesis of fatty acid. 	
b	 i) Outline TCA cycle with its significance. ii) Give the names and structures of substrate and product for the following enzymes: a) Pyruvate dehydrogenase b) Phosphoglycerate kinase c) Lactate dehydrogenase d) Enolase 	18 8 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1
. 2 B	Answer any four questions	48
a	 i) Explain gluconeogenesis with respect to the names of the intermediates and the enzymes involved in reversal of glycolysis. ii) Describe the various complexes involved in ETC. iii) Discuss deamination and decarboxylation reactions involved in amino acid metabolism. 	
b	 i) Give the four steps involved in the β-oxidation of saturated fatty acid. ii) Explain the formation of ketone bodies. iii) Outline the oxidative phase of HMP shunt and give its significance. 	
C C C C C C C C C C C C C C C C C C C	 i) Classify carbohydrates and give two examples of disaccharides. ii) Draw the structures of any two acidic amino acids, and explain α-helix structure of protein. iii) Discuss in brief the steps involved in translation. 	
q	 i) Outline the steps involved in the following a) Synthesis of AMP from IMP b) Salvage pathway for purines ii) Explain in brief about initiation and elongation steps in prokaryotic replication. iii) Define enthalpy and entropy. Discuss the biological role of phospholipids. 	
e	 i) Draw the structure of ATP and explain enzyme induction and repression. ii) Classify enzymes as per IUB system with suitable examples. iii) Discuss Michaelis Menten plot with respect to reversible enzyme inhibitors. 	

UNIVERSITY OF MUMBAI END SEMESTER EXAMINATION AUGUST 2022

Academic Year 2021-2022 First Year B. Pharmacy, Semester II Subject- Environmental Science

Date1/08/2022
Time:. 11.00 to 1.00 pm

Total Marks: 35

Attempt the following question and draw the diagram whenever necessary.

Q:I Attempt **any one** from following questions.

5 Marks

- 1) Define environments. Describe the importance of environment studies. Enlist the renewable and non-renewable resources.
- 2) Comments on Ecological pyramids.

Q: II Attempt **any three** from following questions

30 Marks

- 1. Explain the concept of an ecosystem with their structure and function. Draw an energy flow in the ecosystem.
- 2. Describe are the consequences of air pollution. Explain the impact of air pollution on human health.
- 3. Define the term land degradation and give suitable reasons for the same. Explain preventive measures for the same.
- 4. Define deforestation. Give the reasons for the deforestation. Explain preventive measures for the same.

Subject: Human Anatomy and Physiology II

Duration: 3 Hours

N.B: 1. All questions are compulsory

2. Figures to right indicate full marks

Q1. Choose appropriate option for following multiple choice based questions.	20
1. Neurons secreting dopamine are present in	223623
a. Substantia nigra	22226
b. Arbor vitae	382222
c. Cerebellar peduncles	
d. Pontine nucleus	22000
2cells of stomach secrete pepsinogen and gastric lipase.	
a. G Cells	26322
b. Chief cells	16 17 16 13
c Parietal cells	
d. Mucous neck cells	
3. Each bronchopulmonary segment of the lung has many small compartments called as	· · · · · · · · · · · · · · · · · · ·
a. Lobules	
b. Alveoli	
c. Respiratory bronchioles	
d. Terminal bronchioles	
4 is the middle and the thickest layer of tissue surrounding each kidney	y .
a. Renal Capsule	
b. Renal Fascia	
c. Adipose Capsule	
d. Renal Medulla	
5. If you drank a litre of water, what effect would this have on the osmotic pressure of you	our blood?
a. Increase in osmotic pressure	
b. Decrease in osmotic pressure	
c. No change in osmotic pressure	
d. Slight increase in osmotic pressure	
6is a large, fluid filled follicle that is ready to rupture and expel its secon	dary oocyte.
a. Corpus luteum	
b. Graafian follicle	
c. Corpus albicans	
d. Ovarian follicles	
7. of hypothalamus serves as relay centre for reflexes related to se	ense of smell.
a. Supraoptic region	
b. Tuberal region	
c. Mammillary region	
d. Preoptic region	
8is the organ that stores, concentrates, and delivers bile into duodenum	via common
bile duct	

F.Y.B.Pharm, Sem II

Total marks: 80M

a. Liver
b. Gall bladder
c. Pancreas
d. Large intestine
9 functions in warming, moistening, and filtering air; receives olfactory stimuli; is a
resonating chamber for sound.
a. Nose
b. Pharynx
c. Larynx
d. Epiglottis
10. Which of the following is the correct sequence in which urine flows through the kidney toward the
urinary bladder?
a. Renal pelvis, major calyx, minor calyx, papillary duct, ureter
b. Papillary duct, minor calyx, major calyx, renal pelvis, ureter
c. Minor calyx, major calyx, papillary duct, renal pelvis, ureter.
d. Papillary duct, major calyx, minor calyx, ureter, renal pelvis.
11 are involved in formation of blood testis barrier.
a. Spermatogenic cells
b. Sertoli cells
c. Primordial cells
d. Germinal epithelial cells
12. Which cells produce parathyroid hormone (PTH).
a. Chief cells
b. Oxyphil cells
c. Follicular cells
d. Parafollicular cells
\$\tag{\tag{\tag{\tag{\tag{\tag{\tag{
13. During the phase, the negative membrane potential becomes less negative, reaches
zero, and then becomes positive.
a. Repolarizing
b. Depolarizing
c. Threshold
d. Hyperpolarizing
14 contribute to sperm motility and viability and may stimulate smooth muscle
contractions in the female reproductive tract.
a. Fructose
b. Prostaglandins
c. Clotting proteins
d. Prostate-specific antigen
15. How much is Inspiratory reserve volume in an average adult male?
a. 3100 mL
b. 2100 mL
c. 1100 mL
d. 1500 mL
7 A V 3 D 4 AV 7 AV 7 E 7 V 1 B AV 8 D

16	are the largest and most numerous neuroglia in the CNS.
a. Olig	odendrocytes
b. Astr	ocytes Script Sc
-	ndymal cells
d. Mici	roglia
17. The	e anterior pituitary (anterior lobe), is also called as
	nohypophysis
	rohypophysis
c. Pars	nervosa
d. Pars	intermedia
18. Sec	cretion of Human growth hormone is inhibited by
a. FSH	
b. TRH	
c. GHR	H
d. Som	atostatin School
19	is the small molecule inhibitory neurotransmitter in the CNS.
a. Glut	amate September 1997 1997 1998 1998 1998 1998 1998 1998
b. Aspa	artate SSANCE ENTRY ON EAST SENTE
c. GAB	
d. Subs	stance P
20. A p	portal triad is composed of
a. bile	canaliculi, branch of the hepatic artery and branch of the hepatic vein
b. bile	duct, branch of the hepatic artery and branch of the hepatic vein
	duct, hepatic sinusoid and branch of the hepatic vein
d. cent	ral vein, branch of the hepatic artery and branch of the hepatic vein
Q2 A.	Answer any ONE question.
a.	Explain the formation, composition, and functions of Cerebrospinal fluid, add a note on spinal
A	meninges.
b.	Define Pulmonary ventilation, explain in detail inhalation and exhalation and factors affecting
2000	pulmonary ventilation
Q2 B.	Answer any FOUR questions. 48
) (a.)	Explain generation of action potential, and describe circulation of Cerebrospinal fluid.
b.	With the help of neat labelled diagram explain the anatomy and histology of small intestine.
\$ 5 C.	i. Write a note on regulation of respiratory centres.
S CALL	ii. Explain in detail the blood supply to the kidney and write a short note on ureters.
d.	Draw a neat labelled diagram of pituitary gland. Write a note on anterior pituitary cells and
5,000	their hormones. Draw a next labelled diagram of histology of every Explain the process of cogenesis.
1 1 0 0 0 1	THE THE THE PROOF TO BE A CONTRACTED AT DISTALABLE AT AVAILABLE FOR THE ARCCCC AT ARCHARACIC

First Year B. Pharmacy (SEM-II)(Choice Based) (R-2019)

Duration 3 hours Total Marks: 80

N.B. 1. All questions are compulsory

Subject: Pathophysiology

2. Figures to the right indicate full marks

Q.1.	Choose appropriate option for following multiple choice based questions	20
1	Clearance of injurious stimuli with replacement of injured cells by normal	200
	cells and resuming of normal function is called as	SALA
a	Resolution of Injury	
b	Fibrosis	S S N
С	Necrosis Section 1997	
d	Apoptosis	
2	A purulent exudate, is an inflammatory exudate rich in leukocytes, cell	
	debris and, in many cases, microbes is called as	
a	lymph SESSIFICATION SESSIFICAT	
b	Transudate	
С	Plural Fluid	
d	Pus Park Park Park Park Park Park Park Park	
3	if not scavenged by antioxidant system are harmful to the cell and can lead to cell death.	
a	Adenosine Tri Phosphate	
(S) b)	Reactive Oxygen Species	
6	Saturated Lipids	
0000	Macromolecular Proteins	
4	Formation of new blood vessels during repair phase is called as	
a	Neovascularization	
b	Necrosis	
C. C. C.	Apoptosis	
	Resolution of the injury	
5000	is irreversible necrosis of heart muscle secondary to prolonged ischemia.	
a a	Acute Myocardial Infarction	
67 27 b	Hypertension	
1,00,00	IA V 10K 40' 0'	

Atheroscierosis	
According to the degree of thickness of ventricular wall involved in	
	866
	50000
	200
	80,00
Which of the following does not cause airway narrowing in an asthma	1,68
attack	2016
Destruction of airways	SA AS
	C. No.
A syndrome in which glomerular filtration declines suddenly and is usually	
reversible is called as	
Renal Calculi	
Chronic Renal Failure	
Transient Renal Failure	
Acute Renal Failure	
Polydipsia, polyuria, glucosuria and polyphagia are the clinical signs and	
symptoms of	
Asthma	
Angina pectoris	
Urinary tract infection	
Diabetes Mellitus	
In there is damage to substantia nigra and globus pallidus.	
Parkinson's disease	
Epilepsy	
Gout	
Peptic Ulcer	
Select the appropriate cause of peptic ulcer from the following	
Helicobacter pylori	
Bicarbonate	
Prostaglandin	
Mucosal blood flow	
Acute cerebral ischemic stroke is characterized by the	
	Destruction of airways Mucus hypersecretion Airway edema Bronchospasm A syndrome in which glomerular filtration declines suddenly and is usually reversible is called as Renal Calculi Chronic Renal Failure Transient Renal Failure Acute Renal Failure Polydipsia, polyuria, glucosuria and polyphagia are the clinical signs and symptoms of Asthma Angina pectoris Urinary tract infection Diabetes Mellitus In there is damage to substantia nigra and globus pallidus. Parkinson's disease Epilepsy Gout Peptic Ulcer Select the appropriate cause of peptic ulcer from the following Helicobacter pylori Bicarbonate Prostaglandin Mucosal blood flow

b sudden decrease in blood pressure c sudden increase in brain temperature d sudden loss of blood circulation to an area of the brain 13 Jaundice in sickle cell anaemia can be classified under a Hepatocellular b Congenital hyperbilirubinemia c Haemolytic d Obstructive 14 A person who suffers from ulcerative colitis for a long period of time is at risk for a Colon cancer b Haemorrhoids c Faecal incontinence d Parkinson's disease 15 Which of these definitions describe osteoporosis? a Slow progressive degeneration of the articular cartilage b Inheritable disorder of connective tissue which affects the skeleton, joints, ligaments, teeth, sclerae and skin. c Reduction in bone mass resulting in increased porosity of skeleton d Chronic condition of disordered bone remodelling 16 Which is correct regarding IBD? a Toxic megacolon occurs Crohn's and Ulcerative colitis b Risk of developing ulcerative colitis is higher in smokers than non-smokers c Cobblestone appearance on bowel wall is more characteristic of Crohn disease d Patients with Crohn disease are more at a risk of colorectal cancer than UC patients 17 Which of following organ mainly affected in tuberculosis? a Hypothalamus b Heart		97,77,48,64,79,00°	
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ligaments, teeth, sclerae and skin. c Reduction in bone mass resulting in increased porosity of skeleton d Chronic condition of disordered bone remodelling 16 Which is correct regarding IBD? a Toxic megacolon occurs Crohn's and Ulcerative colitis b Risk of developing ulcerative colitis is higher in smokers than non-smokers c Cobblestone appearance on bowel wall is more characteristic of Crohn disease d Patients with Crohn disease are more at a risk of colorectal cancer than UC patients 17 Which of following organ mainly affected in tuberculosis? a Hypothalamus b Heart	а	Slow progressive degeneration of the articular cartilage	
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patients 17 Which of following organ mainly affected in tuberculosis? a Hypothalamus b Heart	2226	D.O. IN (S. 10.1%) O.W. VO. W. VO. W. V. W. V. V. V. O. V. V. (O.	
a Hypothalamus b Heart	d d	5° , 6' , 5° , 6' , 6' , 6' , 6' , 6' , 6' , 6' , 6	
b Heart SSSSSS	17	Which of following organ mainly affected in tuberculosis?	
b Heart SSSSSS	6 72 V 43	N 19 0 0 0 0 19 N 19 9 N 19 10 19 1	
C Pancreas 8 0 0 0 0	b.	Heart 88888	
		Pancreas	
d Lungs		Lungs	
18 Which of the following disease spread by contaminated food and water?	18	Which of the following disease spread by contaminated food and water?	
a Addison's disease			
b AIDS	60 00 00 pc	AIDS	
C Typhoid	0000 C	Typhoid	

d	Pancreatitis	
	1 and cadds	
19	Which procedure is most likely used to do test for meningitis?	
а	Spinal tap	
b	Throat culture	
С	Tuberculosis skin test	6,00
d	urine test	
20	Syphilis is a complicated and organized sexually transmitted diseases where the causative agent is	11.88 11.89 11.89
а	Moulds Section 1	PON
b	Virus STATES STA	SANS
С	Fungi	
d	Spirochetes	(c)
Q.2 A	Answer any one question	12
a	What are the risk factors for ischemic heart disease? Discuss pathophysiology of myocardial infarction.	
b	What are neurodegenerative disorders? Explain the clinical features and pathophysiology of Parkinson's disease.	
Q.2 B	Answer any four question	48
a	Write a short note on the reactions of blood vessels in acute inflammation and explain the mechanisms of increased vascular permeability.	
00000000000000000000000000000000000000	Explain the signs and symptoms, etiology and pathophysiology of chronic bronchitis.	
	What is hyperthyroidism? Explain the clinical features and etiopathogenesis of hyperthyroidism.	
d	Enlist types of Jaundice. What are the symptoms and pathogenesis of jaundice?	
	Discuss causes, types and pathogenesis of urinary tract infections.	
00.00		

SET II

Subject: Pharmaceutical Organic Chemistry I Year and Sem: First Year SEM-II

Duration: 3 Hours Total marks: 80

Syllabus: CBCS R-2019

N.B.: 1. All questions are compulsory

2. Figures to right indicate full marks

Choose appropriate option for following multiple choice-based questions.	20
Which compound is a positional isomer of 1-chloropentane?	556
	3
CH(CH3)2CHClCH3	
Choose the incorrect option regarding isomerism	
22 6 6 6 6 6 6 6 6 6 6 6 6 6 6 6 6 6 6	
What is the IUPAC Name for the following compound?	
СH ₃	
1-Methyl-2-oxobutanal	
2-Methyl-3-oxobutanal	
The reactivity order of alkyl halides in SN2 mechanisms is	
1° > 2° > 3°	
3°>1°>2°	
3°>2°>1°°	
Low concentration of nucleophile favours	
CNO reaction	
SN2 reaction	
SN2 reaction SN1 reaction Both SN1 and SN2 reaction	
	Which compound is a positional isomer of 1-chloropentane? CH3CH2CHCH2CH2CI CH3CHCH3CH2CH2CI CH2CI C(CH3)3CI CH(CH3)2CHCICH3 Choose the incorrect option regarding isomerism They differ in both physical and chemical properties They have different molecular formula There are two types of isomerism- structural and stereoisomerism Geometric and optical isomerism are two types of stereoisomerism. What is the IUPAC Name for the following compound? CH3 1-Methyl-2-oxobutanal 2-Methyl-3-oxobutanal 2-Methyl-3-oxobutanal 2-Methyl-3-oxobutanal The reactivity order of alkyl halides in SN2 mechanisms is 1° > 2° > 3° 2° > 1° > 2° 3° > 2° > 1° 2° 3° > 2° > 1° Low concentration of nucleophile favours

	~\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\
6	SN2 mechanism proceeds through the intervention of
a	Free radicals
b	Carbonium ion
C	Transition state
d	Carbanion
7	The most reactive alkyl halide is
a	C2H5F
b	C2H5Br
С	C2H5I
d	C2H5Cl
8	An ideal solvent for SN1 reaction -
a	Polar protic solvent
b	Polar aprotic solvent
С	Non polar solvent
d	Levelling solvent
9	Chlorination of methane to give CCl4 is an example of
a	electrophilic addition
b	free radical substitution
С	nucleophilic addition
d	electrophilic substitution
10	Paraffin waxes are graded by its
a	melting point
b	specific gravity
C	ductility
d	viscosity
3	
11	In the addition of HX to a double bond the hydrogen goes to the carbon that already
200	has more hydrogen is a statement of
a	Hund's rule
b	Markovnikoff's rule
C	Saytzeffs rule
d	Anti Markovniknov's rule
50	
12	Which of the following reacts with HBr in presence of a peroxide to give anti
	Markovnikoff's product
a	1-butene
b	2,3 dimethyl 2 butene
C ₂	2-butene
d	3 hexene
13	Why tertiary carbonium ion is more stable than primary and secondary carbonium
	ion
77	due to presence of +I effect

b	due to presence of-I effect
C	due to steric hindrance
d	Both a) and c)
u	
14	Which of the following alkenes will give a mixture of acetone and acetaldehyde on
17	ozonolysis?
0	1 butene
a b	
	2 methyl 2 butene
<u>C</u>	2 butene
d	2 methyl propene
4.5	
15	Which of the following compound is more stable?
a	1,3 butadiene
b	1,4 pentadiene
С	1,5 hexadiene
d	1,2 propadiene
	\$\times \text{2} \tex
16	1,3 butadiene reacts with bromine to mainly give
a	3,4 dibromo 1 butene
b	4 bromo 1 butene
С	1,4 dibromo 2 butene
d	1 bromo 2 butene
17	Which of the following statements is in accordance with Saytzeff's rule?
a	2-Butene is less stable than 1-Butene
b	2,3-Dimethyl-2-butene is more stable than 1-Butene
C	2-Butene is more stable than 2,3-Dimethyl-2-butene
d	2-Methyl-1-butene is more stable than 2,3-Dimethyl-2-butene
u	2-Methyr-1-butene is more stable than 2,3-Dimethyr-2-butene
18	Select the appropriate product for the following reaction.
10	\$ (\$\lambda \rightarrow \tau \rightarrow
15°	
72.60	
2000	
	C_2H_5MgBr ?
6225	
a	butan-2-ol
5 b	Isobutyric acid San
N C	Propionic acid
d	butan-3-ol
19	Identify the product when benzaldehyde reacted with concentrated potassium
3,000	hydroxide.
2003	Benzyl alcohol
b b	
7 40 61	Benzyl alcohol and potassium salt of benzoic acid
(*) ·C.	Potassium salt of benzoic acid
SAC.	Benzoic acid
2 × 2 × 3	
20	What is the name of final addition product when alcohols are added to ketones?

a	Hemiacetal	
b	Acetal	
C	Hemiketals	70,43
d	Ketals	
		320
Q. 2 A	Answer any one question.	12
a	 Explain in details Electrophilic addition of HBr to 1-propene. Write complete reaction, Give the detailed mechanism for addition as per Markovnikoff's rule and the addition in presence of peroxide. Comment on the stabilities of intermediates and products. Give the products obtained on reaction of 2-chlorobutane and alcoholic KOH. Describe the mechanism for formation of both the products. Comment on major and minor product formation. Justify your answer. 	6
b	 Give Reasons: Why hydrolysis of ethyl chloride is SN2 while hydrolysis of tert-butyl chloride is SN1 reaction? Why alkyl chlorides react easily with -OH ions while vinyl halide does with difficulty? Arrange the following in the order of their increasing reactivity towards nucleophilic substitution reaction. CH3F, CH3Cl, CH3I, CH3Br. Justify your answer. Give detailed mechanism for SN1 Reaction. 	6
		l .
Q. 2 B	Answer any four questions	48
_	Answer any four questions I) Write the IUPAC names for the following $\begin{array}{ccccccccccccccccccccccccccccccccccc$	6

		3 1 X.
	III) Draw the tautomeric forms of acetone and identify the tautomeric system.	2
b	I) Compound (A) having molecular formula C3H7NO, on acid hydrolysis gives an acid(B) and ammonia. When (A) treated with bromine and alkali, a compound (C) is obtained which on treatment with nitrous acid yield an alcohol (D) and nitrogen. Compound (C) on reaction with chloroform and alkali produce evil smell due to formation of compound (E). Write structures for all the compounds mentioned by giving appropriate reactions	600000000000000000000000000000000000000
	II) Depict the detailed mechanism for any two:	25
	1) Benzoin condensation	6
	2) Formation of acetal from acetaldehyde	,
	3) Perkin condensation	
c	I) Arrange the following bases in increasing order of basicity, in both solution	6
C	phase and gaseous phase. Justify the order:	
	N-Methylethanamine, Ethanamine and Trimethylamine.	
	II) With the help of reactions Give Qualitative tests for carboxylic acids. Write	6
	the structure and use of tartaric acid and acetyl salicylic acid	
d	I) How will you distinguish primary, secondary and tertiary alcohols by Lucas test?	6
	State chemical reactions. Write the mechanism of acidic dehydration of alcohols.	
		6
	II) Discuss in detail halogenation of alkanes with example. Give use of paraffin.	
e	I) Discuss in detail the general reaction mechanism of nucleophilic addition	6
	reaction for acetal formation.	
	Give the products for the following:	
	1) Acetone +Phenylhydrazine	
95	2) Acetaldehyde + semicarbazide	
26/2/2	3) Benzaldehyde + 2,4 dinitrophenyl hydrazine	
5,55	II) Elaborate on structural Isomerism in organic compounds with examples each	3
6,8,0		
2016	III) Attempt the following conversions.	3
STAD	1) 2-Bromo-2-methyl propane to 2,2-dimethylpropanamide	
63.47	2) Propiononitrile to Methyl propionate	
	3) Acetone to Isopropylmethylamine	
10,000		