

(Time: 3 Hours)

[Total marks: 70]

- N.B: (1) **All** questions are **Compulsory**  
 (2) Answer **all** sub questions **together**.  
 (3) **Draw neat labeled diagram wherever necessary.**

1. (a) Answer the following 12
- i. Give location and function of Areolar connective tissue
  - ii. What is pinocytosis
  - iii. Enlist functions of lymphatic system.
  - iv. Enlist the cardinal signs of Inflammation
  - v. Write the classification of muscles.
  - vi. What is erythroblastosis fetalis
- (b) **Answer the following** 3
- i. Deficiency of which nutrients cause megaloblastic anemia.
  - ii. Holding a book with stretched hand in steady position is \_\_\_\_\_ type of contraction.
  - iii. Enlist various cellular organelles.
2. (a) Answer any **TWO** of the following. 8
- i. Write a short note on Life Cycle of RBC.
  - ii. Classify White Blood Cells (WBCs). Name the respective conditions in which Neutrophil and Eosinophil count increases.
  - iii. Explain in brief Thrombocytopenia and leucopenia.
- (b) Answer any **ONE** of the following. 3
- i. Write a note on Leukemia
  - ii. Explain pathophysiology of megaloblastic and pernicious anemia.
3. (a) Answer any **TWO** of the following. 8
- i. Write a note on cardiac muscles.
  - ii. Describe in detail Neuromuscular Junction
  - iii. Distinguish between skeletal muscle and smooth muscle.
- (b) Answer any **ONE** of the following 3
- i. Draw a neat, labeled diagram showing organization of skeletal muscle.
  - ii. Write a note on isotonic and isometric contractions.
4. (a) Answer any **ONE** of the following. 4
- i. Discuss anatomy and functions of spleen
  - ii. Explain the process of formation and flow of lymph.
- (b) Answer any **ONE** of the following 4
- i. Explain the term Auto-immune disease. Give pathophysiology of Rheumatoid arthritis.
  - ii. Write a note on Myasthenia Gravis.

- (c) Answer any **ONE** of the following 3
- i. Draw a neat labeled diagram of Stratified squamous epithelium, discuss their location and functions
  - ii. Write a short note on Extracellular Matrix of connective tissue.

5. (a) Answer any **ONE** of the following. 4
- i. Write a detail account on passive transport mechanism across membrane.
  - ii. Explain the process of osmosis.

- (b) Answer any **ONE** of the following 4
- i. Give location and function of different types of muscles.
  - ii. Explain following terms
    - a. Neuromuscular Junction
    - b. Muscle tone

- (c) Answer any **ONE** of the following 3
- i. Compare and Contrast between Acute and Chronic type of Inflammation.
  - ii. Explain in brief the process of tissue repair.

6. (a) Answer any **TWO** of the following. 8
- i. Explain terms
    - a) Thrombocytopenia
    - b) Leukemia
    - c) Polycythemia vera
    - d) Anaemia
  - ii. Explain the process of synthesis of RBCs.
  - iii. Enlist functions of various components of blood.

- (b) Answer any **ONE** of the following 3
- i. Discuss various vascular changes during inflammation.
  - ii. Define inflammation, phagocytosis, and exudate

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[Time: 3 Hours]

[ Marks:70]

Please check whether you have got the right question paper.

- N.B:
1. **All** questions are **compulsory**.
  2. **Draw** neat diagrams wherever necessary.
  3. **Figures** to the **right** indicate **full marks**.

1.
  - i) Justify the need for use of renewable sources of energy. **2**
  - ii) Explain the need of environmental education. **2**
  - iii) Give two benefits of Green buildings. **2**
  - iv) Explain secondary air pollutants with examples. **2**
  - v) What do you mean by vermicomposting? **2**
  - vi) Enlist two non-renewable sources of energy. **1**
  - vii) Define: Photochemical smog **1**
  - viii) Explain the term Environmental Impact Assessment(EIA) **1**
  - ix) Define examples of indoor air pollutants with its effect on human body. **1**
  - x) Define Carbon credit. **1**
  
2.
  - i) Write the causes and effects of depletion of soil resources and measures undertaken for prevention of same. **4**
  - ii) Give the roles and duties of Central Pollution Control Board. **4**

**OR**

Give the roles and duties of State Pollution Control Board.

  - iii) Why is depletion of ozone layer a major environmental hazard. **3**
  
3.
  - i) Describe the types of ecosystem and functional components of ecosystem alongwith examples. **4**
  - ii) What is wind energy and explain the working of hydropower generation **4**

**OR**

What is geothermal energy? What are its forms and enlist its applications.

  - iii) Describe the need of public awareness for environment **3**
  
4.
  - i) Write short note on causes and effects of acid rain **4**
  - ii) Give reason. Indoor air pollution is minimized by green buildings **2**
  - iii) Describe the control measures for sustainable development **2**
  - iv) Discuss the entire procedure for obtaining environmental clearance **3**

**OR**

Explain "Taj Mahal" as a case study in environmental legislation.

**TURN OVER**

5. i) Explain Minamata disease case study **4**
- OR**
- What are the causes and effects of noise pollution
- ii) Write a note on techniques of disaster management with reference to earthquake **4**
- iii) Explain the resource utilization as per the carrying capacity **3**
6. i) Describe the causes and effects of depleting nature of forests on the environment. **4**
- ii) With a neat labelled diagram explain principle and working of photovoltaic cell. **4**
- OR**
- With a neat labelled diagram explain principle and working of flat plate collector.
- iii) Classify municipal solid waste. Explain incineration as one of the methods of solid waste management. **3**
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(3 hrs)

[Total Marks 70]

**N.B. 1. All Questions are compulsory.****2. Figure to right indicate full marks****Q1. A. Explain the following terms (any five)**

5

1. Polar covalent bond
2. Heterogeneous catalyst
3. Inductive effect
4. Charge transfer complex
5. First order reaction
6. HOMO

**B. Fill in the blanks (any five)**

5

1. Ground state elect electronic configuration for Magnesium is-----
2. Lewis structure for Nitric acid ( $\text{HNO}_3$ ) is -----
3. ----- Orbital shows only one node
4. Tetracyanoethylene is an excellent acceptor, and it forms -----with electron rich systems such as hexamethylbenzene.
5. The formula for calculation of half-life for first order reaction is-----
6. Crown ether is an example of -----

**C. Match the following**

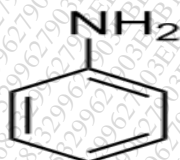
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- |                                |                             |
|--------------------------------|-----------------------------|
| 1. $dx^2-y^2$                  | a) Charge transfer          |
| 2. Carbon in alkane            | b) example of specific base |
| 3. Starch iodine complex       | c) sigma symmetry           |
| 4. $\text{OH}^-$               | d) $3d^7 4s^2$              |
| 5. valence electron of Co (27) | e) $sp^3$                   |

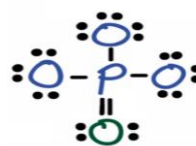
**Q2. A. Draw the resonating structures for**

2

a.



b.



- Q2. B.** Draw Molecular orbital diagram for Ethane. Indicate **HOMO** and **LUMO**. **3**
- C. Fill in the blanks on the basis of Kinetic isotopic effect.** **3**
1. When  $k_H/k_D$  is greater than one, we call the isotope effect ----- and when  $k_H/k_D$  is less than one, we call the isotope effect -----
  2. ----- element shows highest isotope effect
  3. When isotope effect is seen not at rate determining step is called -----
- D.** Define turn over number. Explain metal ion catalysis with example. **3**
- Q3. A.** Give Eyring equation and Arrhenius rate law. Clearly name each term involved in expression **3**
- B.** Calculate rate constant in  $\text{hr}^{-1}$  for the first order reaction with half life of 360 min **2**
- C.** Define group orbital. Mention symmetry elements of  $\text{MH}_3$  system. Enlist molecular orbitals for ammonia
- D.** Compare the energy of linear and bent form of  $\text{MH}_2$  system using molecular orbital diagram **3**
- Q4. A.** Discuss molecular orbital theory **3**
- B.** What do you mean by second order mixing? State any four rules of QMOT. **3**
- C.** Define fast kinetics. Enlist the method to study fast kinetics. Explain any one. **3**
- D.** What is phase transfer catalysis? Give examples. **2**
- Q5. A.** State true or false **3**
- i) Bond length for an alkane is larger than alkene
  - ii)  $\text{KCl}$  is less polarizable than  $\text{NaCl}$
  - iii) Group electronegativity for nitro group is lower than chloro
- B.** Define reaction intermediate. Explain formation of any one. **3**
- C.** A first order reaction was found to have energy of activation of  $2.15 \times 10^4 \text{ J/mol}$ . Calculate the temperature at which reaction will have a rate constant of  $0.030 \text{ sec}^{-1}$ . Frequency factor  $A = 5 \times 10^{13} \text{ sec}^{-1}$  and  $R = 8.314 \text{ J/kmol}$ . **2**
- D.** Write a short note on Charge transfer complexes. **3**

Q6. A. Complete the following table on the basis of hybridization.

3

Molecule	Hybridized state of <u>underlined atom</u>	Bond angle
<u>S</u> F <sub>6</sub>		
<u>C</u> H <sub>2</sub> =CH <sub>2</sub>		
<u>Be</u> F <sub>2</sub>		

B. Write a short note on general acid catalysis.

4

C. Explain Kinetics vs thermodynamics control of reaction with suitable example.

4

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